

NO-Independent Stimulators of Soluble Guanylate Cyclase

Alexander Straub,^{a,*} Johannes-Peter Stasch,^b Cristina Alonso-Alija,^a
Jordi Benet-Buchholz,^c Bernhard Ducke,^b Achim Feuerer^a and Chantal Fürstner^a

^aInstitute of Medicinal Chemistry, Pharma Research Centre, Bayer AG, D-42096 Wuppertal, FRG

^bInstitute of Cardiovascular and Arteriosclerosis Research, Pharma Research Centre, Bayer AG, D-42096 Wuppertal, FRG

^cCentral Research, X-ray Laboratory, Division of Structural Research, Bayer AG, D-51368 Leverkusen, FRG

Received 11 December 2000; accepted 19 January 2001

Abstract—SARs around a novel type of guanylate cyclase stimulator which act by a mechanism different from classical NO-donors are described. Several pyrazolopyridinylpyrimidines are shown to relax aortic rings and revealed a long-lasting blood pressure lowering effect in rats after oral application. © 2001 Elsevier Science Ltd. All rights reserved.

Introduction

Soluble guanylate cyclase (sGC)¹ is a heterodimeric (α/β) heme-protein that converts GTP to cGMP, which is an important messenger in signal transduction. Its natural stimulator is nitric oxide, which stimulates sGC via the formation of a nitrosyl-heme complex. Organic nitrates like glycerol trinitrate or isosorbide dinitrate have been used for decades as a treatment for angina pectoris. In vivo they are converted to NO which relaxes vascular smooth muscles. The major drawback of this therapy is the development of tolerance after repeated applications. Recently an indazole derivative, YC-1, has been described, which stimulates sGC directly via a distinct mechanism and sensitizes the enzyme towards its native activator NO.² We used YC-1 as the lead structure and selected the pyrazolopyridine Bay 41-2272 (**40**) as the most promising one out of a series of thousands of newly synthesized derivatives.

Chemistry³

Pd-catalyzed biaryl-coupling is the method of choice for the preparation of indazole-derived compounds. However this route gives low yields when applied to compounds possessing more *N*-heteroatoms. Therefore, in the case of pyrazolopyridinyl-pyrimidines we applied an

alternative sequence, for which a representative example is shown in Scheme 1.

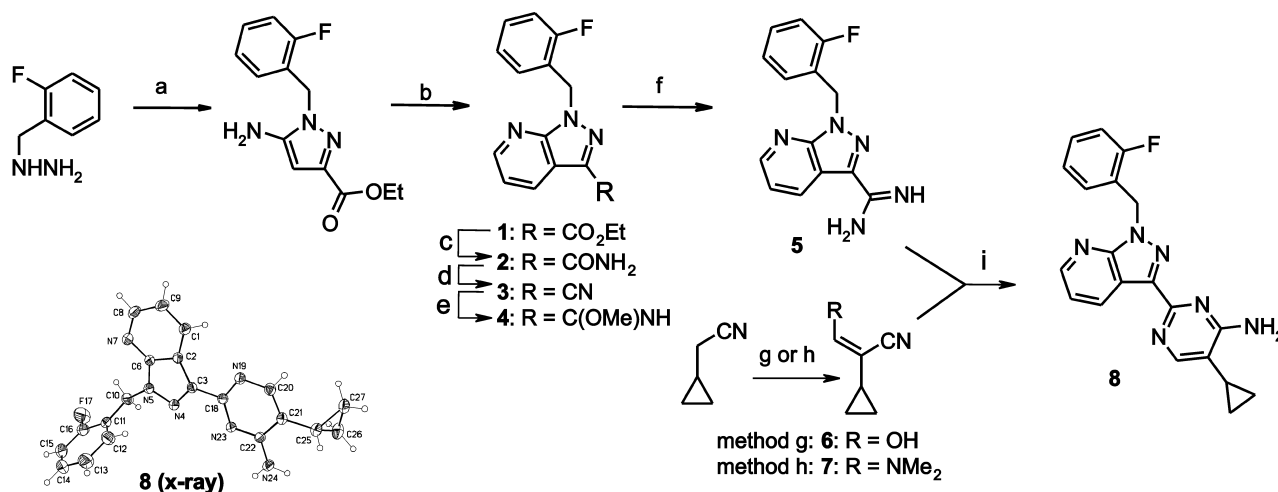
Pharmacology⁴

For biological characterization we measured IC₅₀s for the ability of the compounds to inhibit maximum constriction of phenylephrine-treated precontracted (3×10^{-8} g/mL) rabbit aortic rings. In order to prove the mechanism of relaxation, the stimulatory effect on isolated recombinant sGC alone and in the presence of NO was measured. The in vivo efficacy was shown in anaesthetized rats after oral administration.⁴

Results and Discussion

We started the optimization process with the variation of the benzylic moiety of YC-1. At the methylene bridge no substitution like a carbonyl-oxygen or alkyl groups is tolerated. Omitting the methylene group with directly linked aromatics leads to inactive compounds. Table 1 reflects the influence of substituents at the aromatic part on the relaxation of precontracted aortic rings. Fluorine proved to be preferable leading to compounds which show an increase in activity when changing the position of the substituent from 4 (**3**)→3 (**2**)→2 (**1**). Other halogens were less beneficial. The best activity was found with a 5-pyrimidinylmethyl residue (**10**); however because of its difficult synthetic accessibility and poor solubility we focused on the 2-fluorophenyl moiety for further optimization work.

*Corresponding author at current address. Bayer AG, Agricultural Centre, Bldg. 6550, D-40789 Monheim. Tel.: +49-2173-384972; fax: +49-2173-384880; e-mail: alexander.straub.as@bayer-ag.de



Scheme 1. Reagents and conditions: (a) NC-CH=C(ONa)CO₂Et/TFA, dioxane, reflux, 8 h; (b) Me₂NCH=CH-CHO/TFA/dioxane, reflux, 3 days, 50% for steps (a) and (b); (c) NH₃/MeOH, rt, 2 days, quant.; (d) TFAA/pyridine, pyridine, rt, 8 h, quant.; (e) MeONa/MeOH, rt, 2 h, quant.; (f) 1. NH₄Cl/HAc, MeOH, reflux, 8 h; 2. Na₂CO₃/H₂O, rt, 76.4%; (g) 1. HCO₂Et, *t*-BuOK, THF, rt, 3 h; 2. HCl→pH 4; (h) [Me₂N]₂CHO-*t*-Bu, neat, 100 °C, 46 h, 83.5%; (i) **7**, neat, 100 °C, 12 h, 23.4%; or **6**, toluene, reflux, 8 h, 31.8%.

Concerning the pyrazolo-type central moiety (see Table 2), among all the core heterocycles investigated the pyrazolo[3,4-*b*]pyridin ring system (**12**) seems to be the optimum with respect to *in vitro* effects, whereas indazoles usually are somewhat weaker.

As shown in Table 3, several heterocycles can replace the furyl moiety. Besides other five-membered ring systems like oxazoles (**20–22**), we synthesized a 2-pyridyl derivative (**25**) which had moderate activity. However, after replacing it with a 2-pyrimidinyl-residue (**26**), which displays rotational symmetry with respect to the heteroatoms, a marked increase in potency could be observed. In contrast to the furyl system no further increase in activity was achieved by introduction of a hydroxymethyl group (**27**). However, a 4-amino group significantly enhanced the activity (**28**).

Table 1. Variation of the *N*-substituent

Compounds	R	Relaxation of precontracted aortic rings IC ₅₀ , μM ^a
YC-1	Ph	10
1	2-F-Ph	4.9
2	3-F-Ph	10
3	4-F-Ph	19
4	2,6-F ₂ -Ph	15
5	2-CN-Ph	8.7
6	3-CN-Ph	26
7	2-(5-Chlorothiophene)	7.2
8	2-Pyridyl	16
9	3-Pyridyl	9.6
10	5-Pyrimidinyl	3.2

^aValues are standardized to YC-1 = 10 μM.

We next investigated the SAR for various substituents at the aminopyrimidine (Table 4). A wide diversity of substituents is tolerated in the 5-pyrimidine position without considerably affecting the activity. For alkyl residues, there seems to be an optimum at 2–5 carbon

Table 2. Variation of the pyrazole type core heterocycle

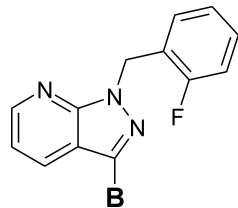
Compounds	Heterocycle A	Relaxation of precontracted aortic rings IC ₅₀ , μM ^a
11		4.9
12		1.7
13		6.932
14		9.8 ^b
15		20.4 (27%) ^c
16		9.4 ^d

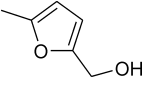
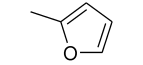
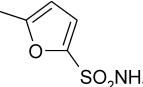
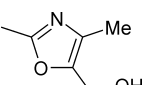
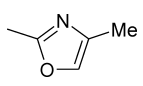
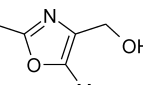
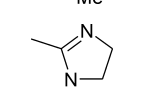
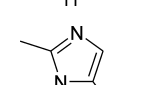
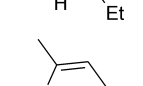
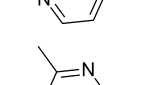
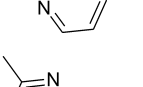
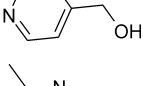
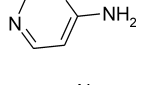
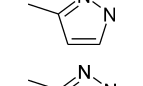
^aValues are standardized to YC-1 = 10 μM.

^bIn combination with 5-hydroxymethyl-4-methyl-1,3-oxazol-2-yl instead of 5-hydroxymethyl-2-furyl (cf. compound **20** in Table 3).

^cPercent of inhibition.

^dIn combination with benzyl instead of 2-fluorobenzyl.

Table 3. Variation of the attached heterocycle


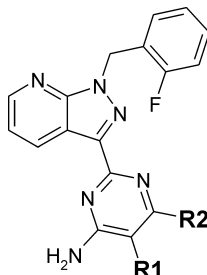
Compounds	B	Relaxation of precontracted aortic rings IC ₅₀ , μM ^a
17		1.7
18		9.3
19		8.0
20		0.9
21		1.2
22		1.5
23		3.2
24		0.6
25		4.8
26		1.25
27		1.3
28		0.27
29		1.8
30		8.6

^aValues are standardized to YC-1 = 10 μM.

chain length. With increasing steric demand and lipophilicity, the activity decreases; the introduction of polar residues generally leads to more potent compounds as compared to alkyl chains with similar steric demand.

We selected **40** for further investigations. The X-ray crystallographic analysis of **40** revealed a coplanar arrangement of the biaryl system (Scheme 1). Since this is a prerequisite for ligand function, we reasoned that these molecules might display metal-complexing properties. Indeed we were able to prepare a silver complex by adding AgNO₃ to **40** in acetonitrile. The X-ray crystal structure is shown in Fig. 1. The relevance of this phenomenon for the biological mechanism is still unclear. However, substituting the core heterocycle by an indole which lacks the ligand nitrogen leads to inactive compounds.

Compound **40** concentration-dependently stimulated isolated sGC from 0.1 nM–100 μM (Fig. 2). When given

Table 4. Variation of the pyrimidine substituents


Compounds	R1	R2	Relaxation of precontracted aortic rings IC ₅₀ (nM) ^a
31	H	H	265
32	Me	H	675
33	Et	H	272
34	<i>n</i> -Propyl	H	252
35	<i>i</i> -Propyl	H	207
36	<i>n</i> -Butyl	H	218
37	<i>t</i> -Butyl	H	709
38	<i>n</i> -Pentyl	H	200
39	<i>n</i> -Hexyl	H	1041
40	Cyclopropyl	H	304
41	Cyclobutyl	H	633
42	Cyclopentyl	H	904
43	1-Cyclopentenyl	H	836
44	Cyclohexyl	H	561
45	CN	H	391
46	2-Cyanoethyl	H	412
47	5-Cyanopentyl	H	588
48	6-Cyanoheptyl	H	1294
49	OMe	H	343
50	SMe	H	243
51	SO ₂ Me	H	261
52	SO ₂ propyl	H	378
53	SO ₂ <i>i</i> -propyl	H	517
54	PO(OEt) ₂	H	414
55	<i>N</i> -Morpholino	H	306
56	F	H	478
57	CH ₂ CF ₃	H	508
58	<i>N</i> -Thiomorpholino	H	650
59	CH ₂ NH ₂	H	1400
60	H	Me	536
61	Cyclopropyl	NH ₂	233

^aValues are standardized to YC-1 = 10 μM.

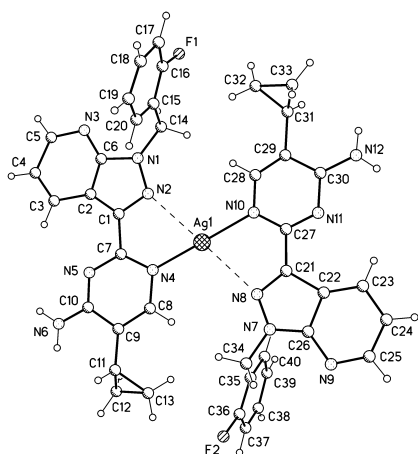


Figure 1. Single crystal X-ray structure of the silver (Ag(I)) complex of **40** (NO_3^- -counterion not shown).

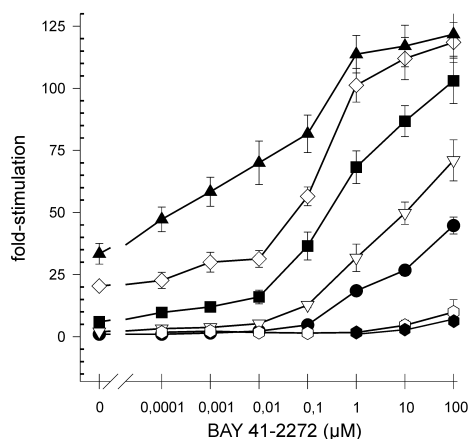


Figure 2. Stimulation of purified sGC by **40** (0.0001–100 μM) in the absence (●) and presence of SIN-1 (▽ 0.01, ■ 0.1, ◇ 1, and ▲ 10 μM) and ODQ (● 10 μM). Stimulation of □ heme-free sGC. The specific activity of sGC is expressed as α -fold stimulation versus basal activity (basal activity in the presence of Mg^{2+} 152 nmol/mg/min). The data presented represent means \pm SEM, from four independent experiments performed in duplicate.

in combination, **40** and the NO donor SIN-1 revealed a strong potentiating effect over a wide range of concentrations. As shown in Fig. 2, **40** does not activate the heme-free enzyme. The sGC inhibitor ODQ completely inhibited the effect of **40** at intact sGC. Investigations concerning the molecular mechanism of action are published elsewhere.⁵ After oral administration to anaesthetized rats it dose-dependently produced a long-lasting decrease in blood pressure (Fig. 3).

Our data demonstrate SARs of NO-independent sGC stimulators which may offer a novel approach for treatment of cardiovascular diseases.

Deviation in mean blood pressure (mm Hg)

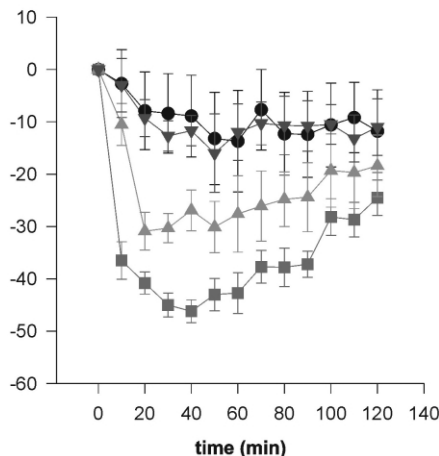


Figure 3. Effect of **40** on blood pressure of anesthetized rats: ● Control ($n=6$); ▼ **40**: 0.3 mg/kg po ($n=4$); ▲ **40**: 1.0 mg/kg po ($n=5$); ■ **40**: 3.0 mg/kg po ($n=4$).

References

- (a) For recent reviews see: Hobbs, A. J. *Trends Pharmacol. Sci.* **1997**, *18*, 484. (b) Moncada, S.; Palmer, R. M.; Higgs, E. A. *Pharmacol. Rev.* **1991**, *43*, 109. (c) Furchgott, R. F. *Angew. Chem., Int. Ed. Engl.* **1999**, *38*, 1870. (d) Murad, F. *Angew. Chem., Int. Ed. Engl.* **1999**, *38*, 1857. (e) Ignarro, L. J. *Angew. Chem., Int. Ed. Engl.* **1999**, *38*, 1882.
- (a) Ko, F. N.; Wu, C.-C.; Kuo, S.-C.; Lee, F.-Y.; Teng, C.-M. *Blood* **1994**, *84*, 4226. (b) Wu, C. C.; Ko, F. N.; Kuo, S. C.; Lee, F. Y.; Teng, C. M. *Br. J. Pharmacol.* **1995**, *116*, 1973. (c) Friebe, A.; Schultz, G.; Koesling, D. *EMBO J.* **1996**, *15*, 6863. (d) Mülsch, A.; Bauersachs, J.; Schäfer, A.; Stasch, J. P.; Kast, R.; Busse, R. *Br. J. Pharmacol.* **1997**, *120*, 681. (e) Becker, E. M.; Wunder, F.; Kast, R.; Robyr, C.; Hoenicka, M.; Gerzer, R.; Schröder, H.; Stasch, J. P. *Nitric Oxide* **1999**, *3*, 55. (f) Hoenicka, M.; Becker, E. M.; Apeler, H.; Sirichoke, T.; Schröder, H.; Gerzer, R.; Stasch, J. P. *J. Mol. Med.* **1999**, *77*, 14.
- (a) All final compounds were characterized by NMR and mass spectrometry. Crystallographic data for **40** and the Ag complex of **40** have been deposited at the Cambridge Crystallographic Data Centre (**40**: CCDC 157014; **40**-complex: CCDC 157015). For experimental details see Straub, A.; Feurer, A.; Alonso-Alija, C.; Stasch, J.-P.; Perzborn, E.; Hütter, J.; Dembowski, K.; Stahl, E. WO 2,000,006,568, 2000. *Chem. Abstr.* **2000**, *132*, 122629. (b) Gordon, D. W. *Synlett* **1998**, 1065. (c) Collot, V.; Dallemagne, P.; Bovy, P. R.; Rault, S. *Tetrahedron* **1999**, *55*, 6917.
- Becker, E. M.; Wunder, F.; Kast, R.; Robyr, C.; Hoenicka, M.; Gerzer, R.; Schröder, H.; Stasch, J.-P. *Nitric Oxide* **1999**, *3*, 55.
- Stasch, J.-P.; Becker, E. M.; Alonso-Alija, C.; Apeler, H.; Dembowski, K.; Feurer, A.; Gerzer, R.; Minuth, T.; Perzborn, E.; Pleiß, U.; Schröder, H.; Schroeder, W.; Stahl, E.; Steinke, W.; Straub, A.; Schramm, M. *Nature* **2001**, *409* (in press).